21-Jun-91 Unnerst Public			
the second se		the set of a set free the loss that the set of	
From :91/05/21:0000 Dopth Range (0 0 0 Sample State: Freeb Water		10 : 91/ 05/2 Tido :	1:0000
Parameter Description		Units	Analytical Technique
$_{ m D}$ H	8.2	pH units	Automated pH Meter
Specific Conductance	270	uS/cm	Cond. Meter Siebold
Turbidity .	0.3	NTU	Nephel Single Beam
Alkalinity Total	129	mg/L	Automated Electrometer
Hardness Total	152 -	mg/L	Calculated Result
Coliform - Total NO COLIFORMS DETECTED.	< 2.0	CFU	Filter-Membrane
Fluoride Dissolved	0.11	ing /L.	Y. S. E.
Boron	0, 02	ing /L	ICP Analysis
Barium	0.04	my/L	ICP Analysis
Calcium	33.5	ma / L	ICP Analysis
Chromium	< 0.01	mg /L	ICP Analysis
Copper	0.06	mg / L	ICP Analysis
Iron	< 0.01	的真才知	ICP Analysis
Magnesium	10.1	mg≠Ĺ	ICP Analysis
Manganese 🎫	< 0.01	mg ∕t_	ICP Analysis
Zinc	0.02	mo∠L.	ICP Analysis
Nitrogen N03+N02	0.09		Auto. Cadmium Reduction
Nitrogen NO3 Dis	0.09	(Pro) + La	Calculated Result
Nitrogen NO2 Dis	< 0.005		Auto. Diazotization

R General Public		TTN: SANDEE	4 RG, SALLY Sample 8800220 ================================	
Site: NOSITE TYEE L	AKE		nar an the same oper year and the and the same the same top and the same the same the same oper same same and t	
From : 88/05/10:0000	To :88/05/10:0000			
Depth Range : 0.0 0.0 Sample State: Fresh Water Sample Comment:		Tide :		
Parameter Description	Result		Analytical Technique	
рН	8.0	pH units	Automated pH Meter	
Specific Conductance	265	uS/cm	Cond. Meter Siebold	
Turbidity	0.2	NTU	Nephel.Single Beam	
Alkalinity Total 4.5	134	mg∕L	Automated Electrometer	
Hardness Total	127	mg/L	Calculated Result	
Coliform - Total	< 2.0	No/100ml	Filter-Membrane Health L	
Fluoride Dissolved	< 0.10	mg/L	I. S. E.	
Boron	0.07	mg/L	ICP Analysis	
Barium	0.04	mg∕L	ICP Analysis	
Calcium	35.0	mg/L	ICP Analysis	
Chromium	0.01	mg/L	ICP Analysis	
Copper	0.06	mg/L	ICP Analysis	
Iron	0.01	mg/L	ICP Analysis	
Magnesium	9.51	mg/L	ICP Analysis	
Manganese	O. 01	mg/L	ICP Analysis	
Zinc	0.03	mg/L	ICP Analysis	
Nitrogen NO3+NO2 Dis	0.12	mg∕L	Auto. Cadmium Reduction	
Nitrogen NO3 Diss(N)	0.12	mg/L	Calculated Result	
Nitrogen NO2 Diss(N)	< 0.005	mg∕L	Auto. Diazotization	

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FEBRUARY 16, 1983 E	ISTRY OF THE	ENVIRONMENT	
WATER QU	JALITY REPORT	FOR SAMPLE 300990W	
TOI	CARL SANDBER	G	
	BOX 248		
A T T E	SMITHERS BC NTION OF: CA	RI SANDBERG	
		KNOWN AS MELURE LK TAP WA	TER
SAMPLING DATE DATE PROCESSED	(S): JAN 31/8 D TO COMPUTER	3 0800 HRS : FEB 01/83	
РН	REL UNIT	RESIDUE: FILTERABLE	164. MG/L
SPECIFIC CONDUCTANCE	278. UMHD/CM	TURBIDITY	N'.T.U.
ALKALINITY: TOTAL	136. MG/L	FLUORIDE	0.10 MG/L
HARDNESS	112. MG/L	NITROGEN: NITRATE (NO3)	0.1* MG/L
NITROGEN: NITRITE (NO2)	L 0.005* MG/L	BORON TOTAL	MG/L
CALCIUM TOTAL	34.4 Mg/L	CHROMIUM TOTAL	L 0.01 MG/L
COPPER TOTAL	C.D6 MG/L	IRON: TOTAL	0.03 MG/L
MAGNESIUM TOTAL	6.28 MG/L	MANGANESE TOTAL	L 0.01 MG/L
ZINC TOTAL	0.1 MG/L	BARIUM TOTAL	0'.03 MG/L
COLIFORMS: TOTAL	N. 9. N.		
	MINISIRT	FORMED BY DIVISION OF LABO	RATORIES,
THE SYMBOL "L" F The symbol "G" F	REPRESENTS LE Represents gr	SS THAN EATER THAN	
		Aande	and the second s
		FOR ENVIRONMENTAL	LABORATORY

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Water Quality Check Program - Guide to the Interpretation of Analyses

The report that you have received from the Environmental Laboratory shows several analytical results. This booklet is intended to help you understand that report. Included in this booklet are the "Canadian Drinking Water Standards" to enable you to compare your results. If your analyses are above the Canadian standards, we recommend that you contact your local Health Unit for further recommendations.

Concentration Levels may vary according to the time of the year (season and climatic conditions) that your sample was taken. Spring run off and fall rains can have an effect on some tests; the turbidity of the water could be high after a rain storm as the rain washes silt and dirt into the streams and increases the turbidity; in the summer, if there is no water flow (stagnant conditions) and the temperature is warm, then conditions could be optimal for bacterial growth. The limits for the tests that we report are:

Canadian Drinking Water Guidelines

<u>Test</u>		Maximum Acceptable Level (see note 1)	Objective Level (see note 2)
1 2 3 4 5 6	pH Conductance Specific Turbidity Alkalinity Tot. 4.5 Hardness Total * Coliform Total	6.5 to 8.5 (no limit set) 5 NTU (see note 3) 30 to 500 mg/L (see note 4) (no limit set - see below) * < 2 per 100 ml (see note 5)	(no limit set) (no limit set) < 1 NTU (see note 3) (no limit set) (no limit set) < 2 per 100 ml
7 8 9 10 11 12 13 14 15 16 17	Fluoride Dissolved Boron Barium Calcium Chromium Copper Iron Magnesium Manganese Zinc Nitrogen:Nitrate +	< 1.5 mg/L 5.0 mg/L 1.0 mg/L (no limit set - see below) * 0.05 mg/L 1.0 mg/L < 0.3 mg/L (no limit set-see below)* 0.05 mg/L 5.0 mg/L 10.0 mg/L	(see note 5) < 1.0 mg/L < 0.01 mg/L < 0.10 mg/L (no limit set) < 0.0002 mg/L < 1.0 mg/L < 0.05 mg/L (no limit set) < 0.01 mg/L < 5.0 mg/L < 0.001 mg/L
18	Nitrite (as N) Nitrogen:Nitrate (as N)	< 10.0 mg/L	< 0.001 mg/L
19	(as N) Nitrogen:Nitrite (as N)	< 1.0 mg/L	< 0.001 mg/L
	 * Hardness * Calcium * Magnesium Reference: 	< 120 mg/L < 200 mg/L < 150 mg/L Water Quality Sourcebook-Inland Wat Directorate, Environment Canada (197	

Notes:

1 Maximum Acceptable Level for health or aesthetic reasons

- 2 Objective Level is a desired level to be attained
- 3 "NTU" is Nephelometric Turbidity Units

4 "mg/L" is milligrams per liter

5 Coliform count per 100 ml sample

The following notes are brief descriptions of the tests and what the results could mean:

1 pH

2

3

Definition: This is the measurement of the hydrogen ion concentration in the water. A pH between 0 and 7 is acidic (the lower the number, the more acidic the water is) and a pH between 7 and 14 is basic (the higher the number, the more basic the water is). A pH of 7 is considered to be neutral.

Importance: The acceptable level for drinking water is between 6.5 and 8.5. Corrosion of metal plumbing may occur at both low and high pH values. "Scaling" or encrustation of metal pipes may also occur at high pH values.

Conductance Specific

Definition: This is the measurement of the ability of the water to conduct an electric current - the greater the content of the ions in the water, the more current the water can carry. Ions are dissolved metals and other dissolved materials. Conductance is reported in terms of microsiemens per centimeter (uS/cm). There is no standard recommended level for this test. Natural waters are found to vary between 50 and 1500 uS/cm.

Importance: Specific Conductance may be used to estimate the total ion concentration of the water.

Turbidity

Definition: This is the measurement of the suspended particulate matter in the water, which interferes with the passage of a beam of light through the water. The material suspended in the water could be silt, clay, organic material, or micro-organisms.

Importance: Turbidity values may be high during and after periods of high rainfall due to the possibility of silt being washed into streams and wells. High levels of turbidity may protect micro-organisms from the effects of attempted disinfection procedures. It may also stimulate the growth of bacteria and thus increase the 'chlorine demand' (bleach) that may be required to disinfect the water.

Alkalinity Total at pH 4.5

Definition: This is the measurement of the water's ability to neutralize acids. It usually indicates the presence of carbonate, bicarbonates, or hydroxides. Alkalinity results are expressed in terms of an equivalent amount of calcium carbonate. Note that this does not mean that calcium carbonate was found in the sample.

Importance: Natural waters rarely have levels that exceed 500
mg/L. A range of 30 to 500 mg/L is acceptable.
However, extreme variations or high levels may cause
gastrointestinal problems in humans.

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